

Question	Answers	Extra information	Mark	AO Spec reference
01.1	HOOC H CH ₃ NH ₂		1	3.3.7, Ms 4.2
01.2	H ₂ N ^{WWWC} COOH H ₃ C	Check that this is not the same isomer drawn rotated	1	3.3.7, MS 4.2
01.3	Two compounds which have the same chemical formula but different spatial arrangement of atoms, and are non super-imposable mirror images of each other.		1	3.3.7
01.4	HN COOH		1	3.3.13.1
01.5	HN COOH		1	3.3.13.1
01.6	$\begin{array}{cccc} CH_3 & H & O & H_3C & OH \\ H_2N & & & & \\ H_2N & & & & \\ O & & H & O \end{array} OH$		1	3.3.13.2, MS 4.2



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02.1	$HO \rightarrow O \rightarrow$		1	3.3.13.2
02.2	Condensation (polymerisation)		1	3.3.13.2
02.3	H = H = H = H = H = H = H = H = H = H =	One mark for each amino acid drawn correctly.	2	3.3.13.2, MS 4.2
02.4	Thin-layer chromatography could be used to separate the amino acids UV light/ninhydrin could be used to highlight/observe the amino acids The amino acids could then be identified using their $R_{\rm f}$ values compared to known values	OWTTE	1 1 1	3.3.13.2
03.1	A compound that contains both an amine group and a carboxylic acid group		1	3.3.13.1
03.2	H ₃ C OH NH ₂ OH	Either chiral centre can be circled.	1	3.3.7, MS 4.2



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03.3	$H_{3}C \underbrace{\bigcirc}_{H_{3}C} OH H_{3}C \underbrace{OH H_{3}C} OH H_{3}C $	Make sure there are no repeats of structures	3	3.3.7, 3.2.5.3, MS 4.2
03.4	$\begin{array}{cccc} CH_3 & H & O & H_3C & OH \\ H_2N & & & & \\ & & & & \\ & & & & \\ O & & & H & O \end{array} OH$		2	3.3.13.1
04.1	Enzymes act as catalysts Provide a pathway with a lower activation energy (and therefore increase the rate of the chemical reaction)		1 1	3.3.13.3
04.2	The drugs can have a similar shape/can fit in to the stereospecific active site of the protein (blocking the active site/acting as a competitor)		1 1	3.3.13.3
05.1	Hydrogen bonds/bonding		1	3.3.13.4
05.2	[DNA strand] N (hydrogen bonds)		2	3.3.13.4, MS 4.2
05.3	ligand substitution / replacement where platinum of cisplatin bonds to N on guanine, (which alters the structure of the DNA preventing replication)		1 1	3.3.13.5
05.4	β-pleated sheets		1	3.3.13.2

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05.5	The primary structure is the order of amino acids.					1	3.3.13.1
06.1	С	Н	N	0	Need a comment about the empirical formula being the molecular formula/having the $M_{\rm r}$ = 103 for the last mark.		3.1.2.4, MS 0.1, MS 0.2, MS 2.2,
	46.6/12.0	8.7/1.0	13.6/14.0	31.1/16.0		1	
	= 3.8	= 8.7	= 0.971	= 1.94		-	
	3.88/0.971	8.7/0.971	0.971/0.971	1.94/0.971		1	
	4	9	1	2			
	Empirical formula = $C_4 H_0 NO_2$				1		
	Which has M_r of 103.0 g mol ⁻¹ so that must be the molecular formula.				1		
06.2	$\begin{array}{c} COOH \\ C_{2}H_{5} \\ H \end{array} \xrightarrow{C} H \\ NH_{2} \\ NH_{2} \\ NH_{2} \\ H \\ C_{2}H_{5} \\ H \\ C_{$			Check that there are two different isomers here, not the same isomer but rotated Allow e.c.f. from 06.1	2	3.3.7, MS 4.1	
06.3	C_2H_5 H C_2H_5 H C_2H_5 H C_2H_5 H C_2H_5 H C_2H_5				Either zwitterion could have been drawn	1	3.3.13.1
06.4	Ion with both a positive charge and a negative charge at the same time					1	3.3.13.1
06.5	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					2	3.3.12.1, MS 4.2
06.6	Condensation					1	3.3.12.1

Skills box answers:









6. For example

